

## Conference Paper

# Reduction of the Negative Impact on the Environment By Optimizing the Combustion Process in Diesel Engines

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**Abstract**

The article considers the problem of the negative impact of the exhaust gases of diesel internal combustion engines on the environment and human health. The types of organization of the ignition process and the process of fuel combustion in a diesel engine are considered. The reasons for the occurrence of increased particulate matter in internal combustion engines in exhaust gases are also described. The main factors affecting the delay of ignition are given. The main stages of soot formation in diesel internal combustion engines are described. The influence of temperature distribution in the jets of injected fuel and the dependence of emissions on the coefficient of excess air are considered. As a result, the main conclusions are given on ensuring the reduction of solid particles in the exhaust gases of diesel engines by optimizing the combustion process.

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## 1. Introduction

The exhaust gases of automotive internal combustion engines have a significant impact on the environment and human health. The issue of reducing particulate matter is particularly relevant in our time. The main harmful components of automobile exhaust gases include:

- carbon monoxide CO (strong toxic substance);
- hydrocarbons CH<sub>x</sub>;
- nitrogen oxides NO<sub>x</sub> (toxic, together with hydrocarbons CH forms photochemical smog);
- aldehydes (harmful effects on the nervous system and respiratory system);
- solid particles, PM (soot);
- sulfur oxides SO<sub>x</sub>;
- benzopyrene;
- lead salts (highly toxic substances) [2].

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Among the named "harmful" components of the exhaust gases, PM 2.5 (BC, "black carbon") should be highlighted. Aircraft emissions are especially high for diesel engines where diffusion combustion takes place. However, almost the same combustion occurs in engines with spark ignition and direct fuel injection. The aircraft emission from such engines is poorly studied and, in our opinion, is an urgent problem that needs to be solved [2].

Methods of reducing atmospheric pollution with exhaust gases of internal combustion engines can be divided into two groups:

- methods for reducing emissions toxicity;
- methods for reducing emissions [3].

## 2. Methods and Equipment

The processes of diesel combustion in internal combustion engines were investigated. During the research, the stages of supplying diesel fuel to the combustion chamber, the formation of a fuel-air mixture and its further ignition were considered.

When organizing the ignition process in diesel internal combustion engines, different methods can be used, such as injection into separate combustion chambers or direct injection. Currently, the use of prechamber and vortex chamber mixture formation is not used so often. Now direct injection systems are used.

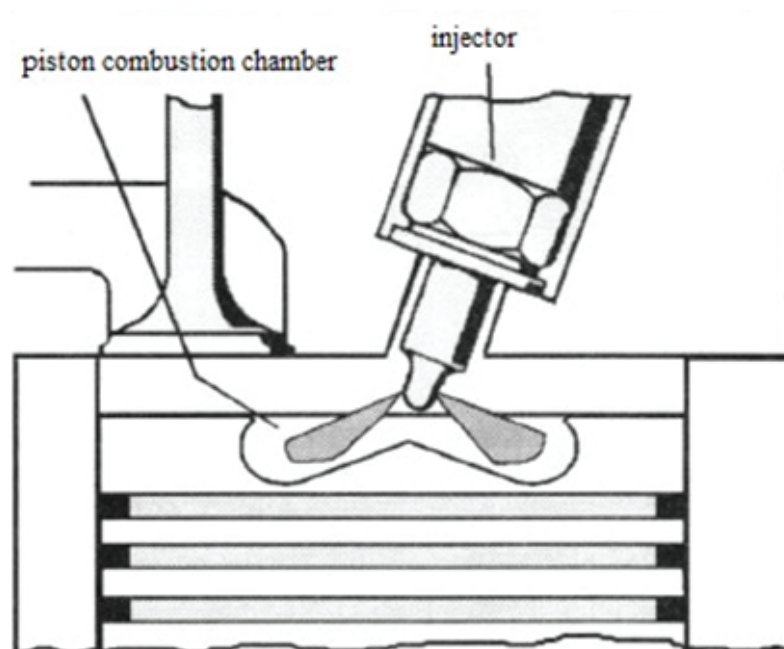


Figure 1: Direct injection.

The main difference between diesel internal combustion engines of trucks and cars is the geometry of the combustion chamber and the types of nozzles used.

The main advantages of direct injection:

- the ability to reduce fuel consumption;
- higher thermal efficiency;
- less heat loss.

Disadvantages of direct injection:

- the vortex flow interferes with the cleaning of the combustion chamber;
- problems with a cold start;
- it is difficult to cool the combustion chamber.

The position and number of jets in the combustion chamber with direct injection affect:

- conditions of mixture formation;
- the process of ignition of the mixture;
- engine efficiency;
- pressure rise rate;
- engine power and torque;
- conditions for soot formation and particle ejection;
- use of air to burn fuel;
- fuel consumption.

### 3. Results

The formation of soot and particulate matter in the exhaust gases of diesel internal combustion engines occur under three basic conditions:

- injection of cold, liquid fuel into the combustion front;
- lack of oxygen and poor conditions for the formation of the mixture;
- drop in pressure or temperature in the combustion chamber and premature termination of combustion.

Ignition retardation is the time from the start of injection to the start of pressure rise above the compression pressure.

The combustion process and the composition of the exhaust gases is determined by the ignition delay process [13].

TABLE 1: Factors Affecting Ignition Delay.

Influencing value	Influence at	Requirements for optimal ignition delay
Compression ratio	- Temperature difference between air and fuel - air pressure and heat transfer coefficient - oxygen concentration in the combustion chamber	Big as possible
Initial compression pressure	- Total pressure in the engine and heat transfer coefficient - oxygen concentration in the combustion chamber	Big
Initial compression temperature	Temperature difference between air and fuel	Big
Engine speed	Time to transfer heat from air to fuel	Small
Engine load	- Level of temperature and pressure in the cylinder - heat transfer coefficient	Big
The movement of air in the combustion chamber	- Heat transfer from air to fuel - speed difference between air and fuel	Very big
Spray quality	- Fuel droplet size - the ratio of the surface area of the droplets to their volume	Very big
Fuel density	The surface area of the fuel droplets at constant weight	Small
Specific heat of fuel	The required amount of heat for heating the fuel and the time for delaying ignition	Small
Flammability	Decay processes of hydrocarbon chains	Cetane number greater than 55

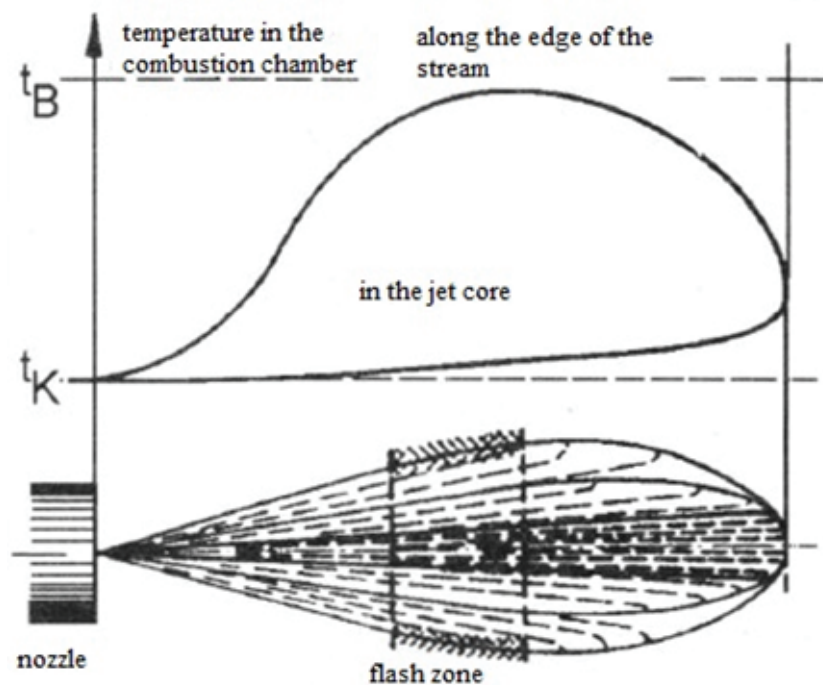
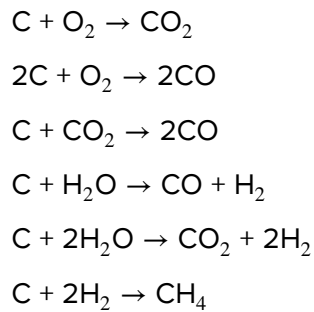
The great importance in preparing fuel for which ignition is presented by the temperature distribution in the jets of injected fuel into the combustion chamber.

Figure 2 shows the temperature distribution in the injected jet when it comes in contact with hot air in the combustion chamber.

The light fuel components located on the periphery of the jet are crushed to even smaller droplets of fuel upon contact with air. The components of the fuel that come into contact with the hot air first boil and evaporate. In the core of the jet, the fuel remains cold, liquid and unprepared for a long time.

Accelerated soot formation is observed at temperatures in the combustion chamber above 2050 K, and its maximum concentration at a temperature of about 2200 K. At higher temperatures, the rate of soot oxidation begins to exceed the rate of soot formation and the amount of soot in the exhaust gases decreases.

The soot oxidation proceeds according to the following main reactions:

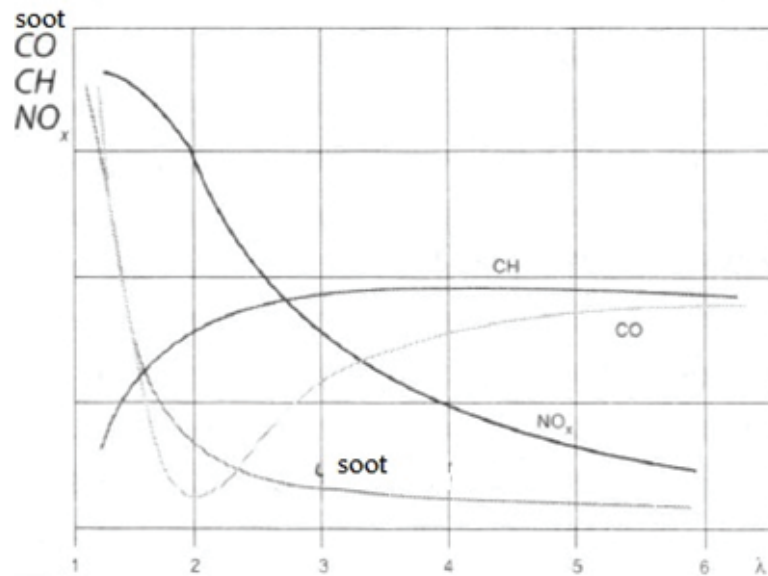


**Figure 2:** Temperature distribution:  $t_K$  -- temperature in the core of the fuel jet,  $t_B$  -- temperature along the edge of the fuel stream

At the same time, a significant part of the soot burns out in the combustion chamber at the expansion stroke and in the exhaust system of the diesel engine.

Primary soot particles have a diameter of about 0.02 - 0.17 microns, and soot in the exhaust gases of diesel engines can be in the form of formations, most of which have a size of 0.5 microns. With an arithmetic average diameter of about 0.3 microns, soot particles have a very developed surface equal to about 90 m<sup>2</sup> per 1 g of carbon black [12].

When soot is inhaled, its particles have a harmful effect on the respiratory organs of a person. They reach the alveoli of the lungs or are deposited in the sinuses, trachea or bronchi. Moreover, large particles of soot (2-10 microns or more) are easily excreted from the body, and small ones (0.5-2 microns) are retained in the lungs, causing chronic diseases. But the main toxic properties of soot are not caused by carbon, but by the presence of carcinogenic polycyclic aromatic hydrocarbons in it, including the most



**Figure 3:** Coefficient of excess air and emissions of harmful substances from a diesel engine: CO - carbon monoxide; CH are hydrocarbons; NO<sub>x</sub> - nitrogen oxides; Soot (including BC).

toxic among them - benz (a) pyrene C<sub>20</sub>H<sub>12</sub>, which is an indicator of the presence of other polycyclic arenes in the exhaust gas.

Particulate matter emissions are mainly affected by the presence of a sufficient mass of oxygen in the formation and combustion of the mixture. In the range close to full load, particulate emissions increase [1].

## 4. Discussion

The researches in the field of completeness of combustion of diesel fuel are promising, as they can help increase the efficiency of internal combustion engines, which in turn will reduce the negative impact on the environment and human health. It is also necessary to search for the most suitable fuel engine for operation in terms of its quality composition.

## 5. Conclusion

For combustion with the formation of the least amount of soot, it is necessary to use a long ignition delay.

In this case, too much fuel can be prepared, which can lead to pressure surges. As a result, more nitrogen oxides NO<sub>x</sub> can form in the exhaust gases. It should also be noted that fuel combustion depends on its quality composition and, accordingly, its ability to efficiently atomize, mix and further ignite it. When the percentage of heavy fractions in

the fuel composition is high, the likelihood of incomplete combustion of hydrocarbons in a diesel engine is greater.

The data from preliminary studies will be used to further study the dependence of the design features of internal combustion engines and the composition of exhaust gases.

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## Conflict of Interest

The authors have no conflict of interest to declare.

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